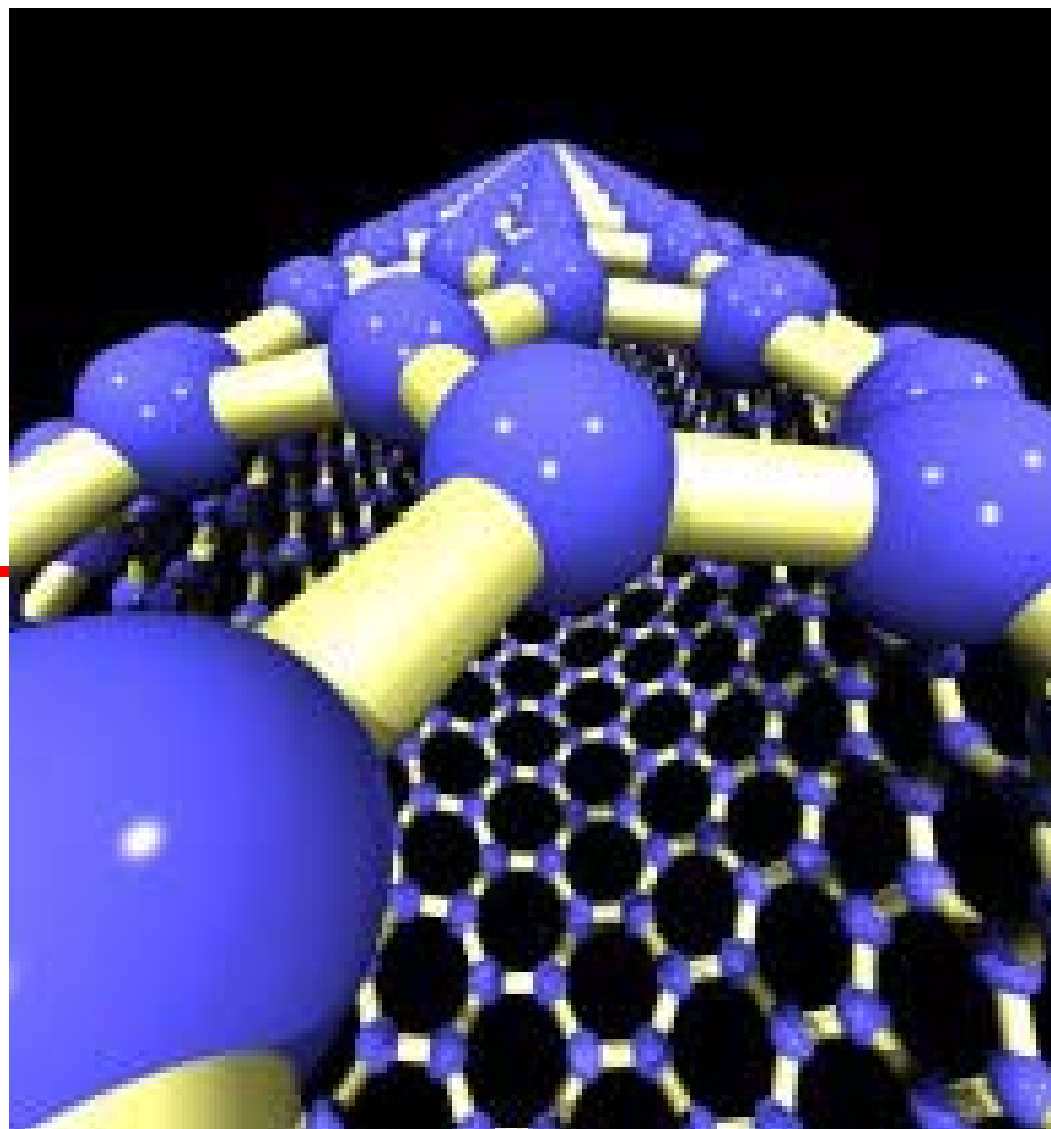


Nanomaterials with Antimicrobial Properties: Mechanisms, Implications and Applications

Synchrotron SOLEIL
Massy, 16 Sep 2009

Pedro J.J. Alvarez



Acknowledgements: NSF/CBEN, EPA

Students

- Del Lyon
- Dong Li
- Katherine Zodrow
- Matt Hotze
- Zongming Xiu

Postdocs

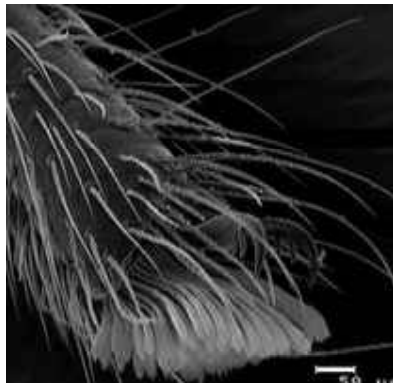
- Jaesang Lee
- Shaily Mahendra
- Lena Brunet

Faculty

- Mark Wiesner
- Qilin Li
- Jaehong Kim
- Vicki Colvin
- Greg Lowry
- Lon Wilson
- Mike Wong
- Joe Hughes

Responsible Nanotechnology

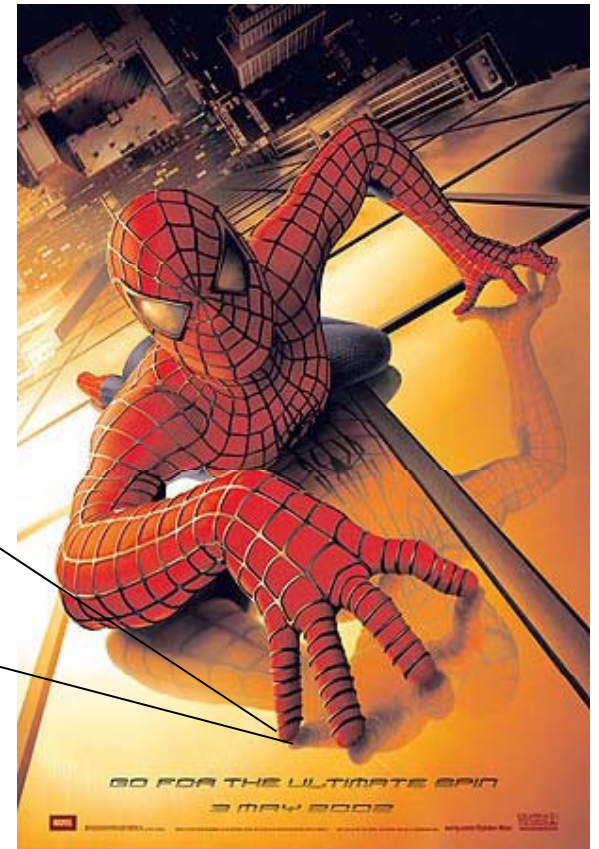
"With Great Power, Comes Great Responsibility"
Uncle Ben to Peter Parker in Spider Man



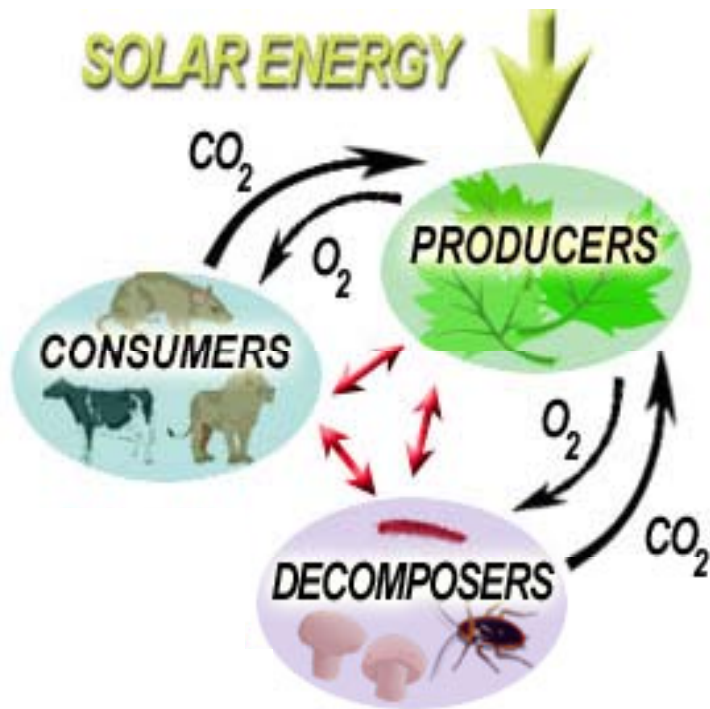
Nanoscale fibers on
spider feet

Paul Hermann Muller
Thomas Midgley

(Dr. Nigel Walker, NIEHS NTP)



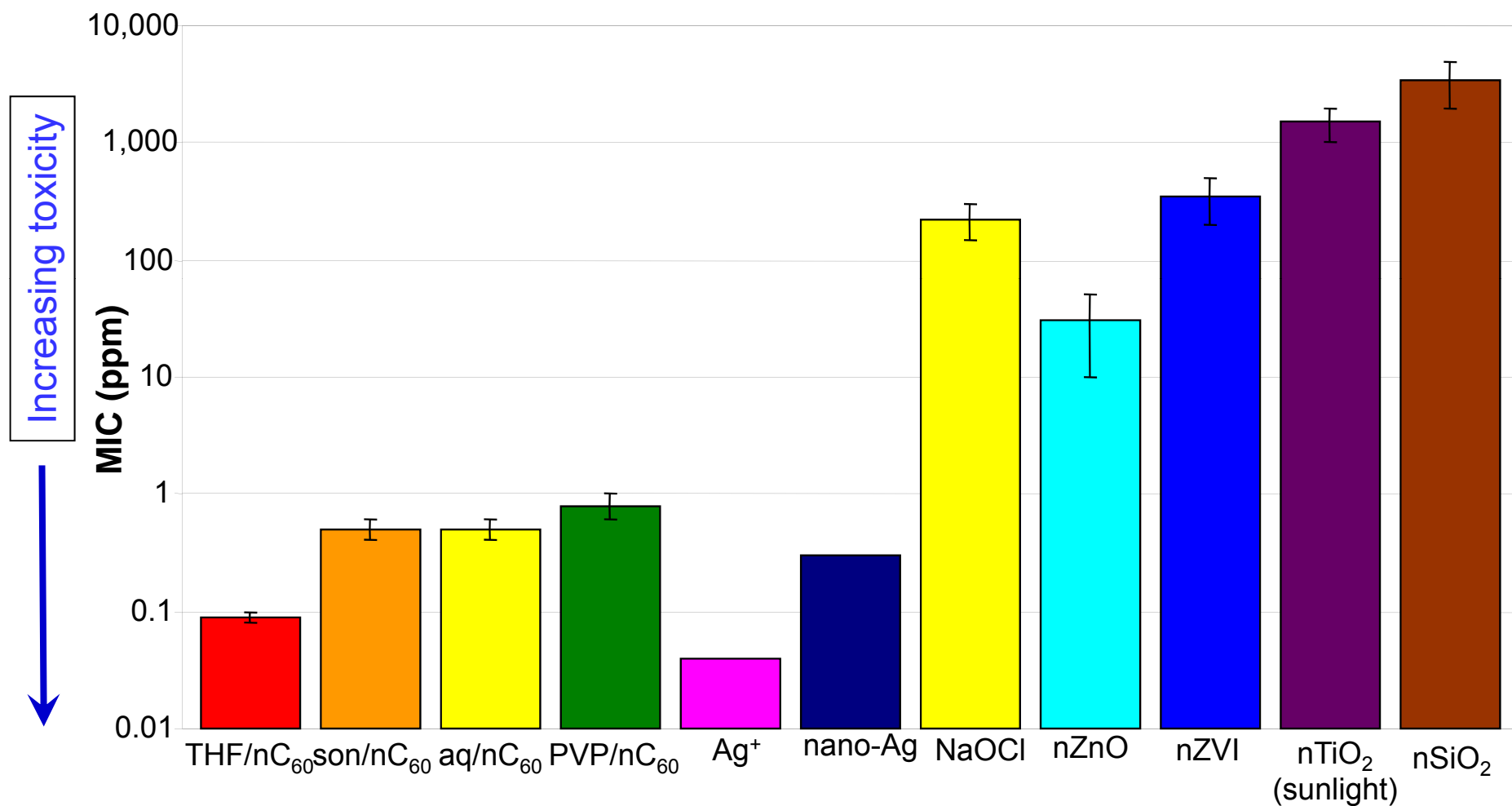
Microbial-nanoparticle Interactions to Inform Risk Assessment



- Bacteria are at the foundation of all ecosystems, and carry out many ecosystem services
- Disposal/discharge can disrupt primary productivity, nutrient cycles, biodegradation, agriculture, etc.
- Antibacterial activity may be indicative of toxicity to higher level organisms



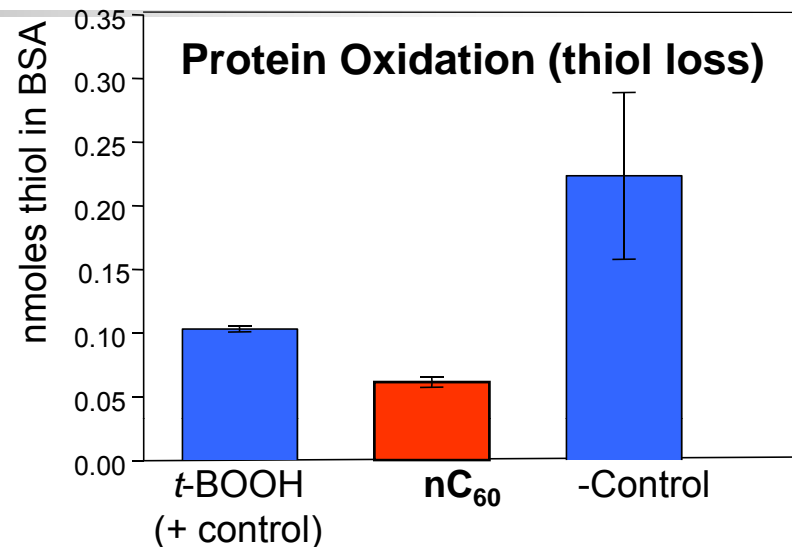
Toxicity of common nanomaterials to *B. subtilis*



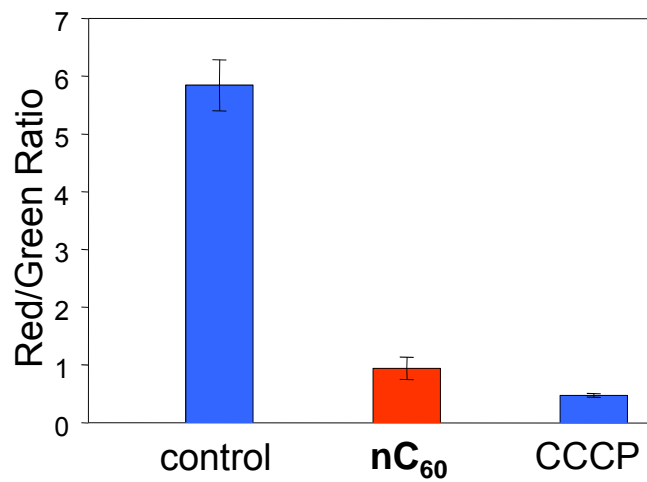


Antibacterial Mechanisms of nC₆₀

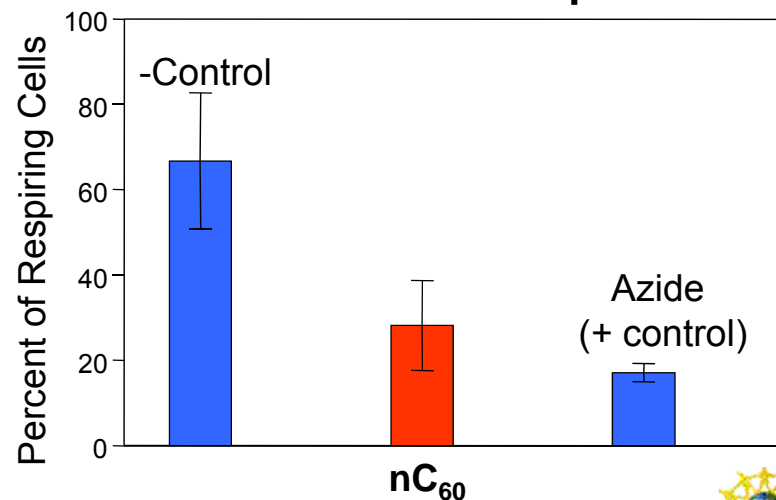
nC₆₀ exerts ROS-independent oxidative stress, with evidence of protein oxidation, collapse of membrane potential, and *interruption of cellular respiration & energy transduction*



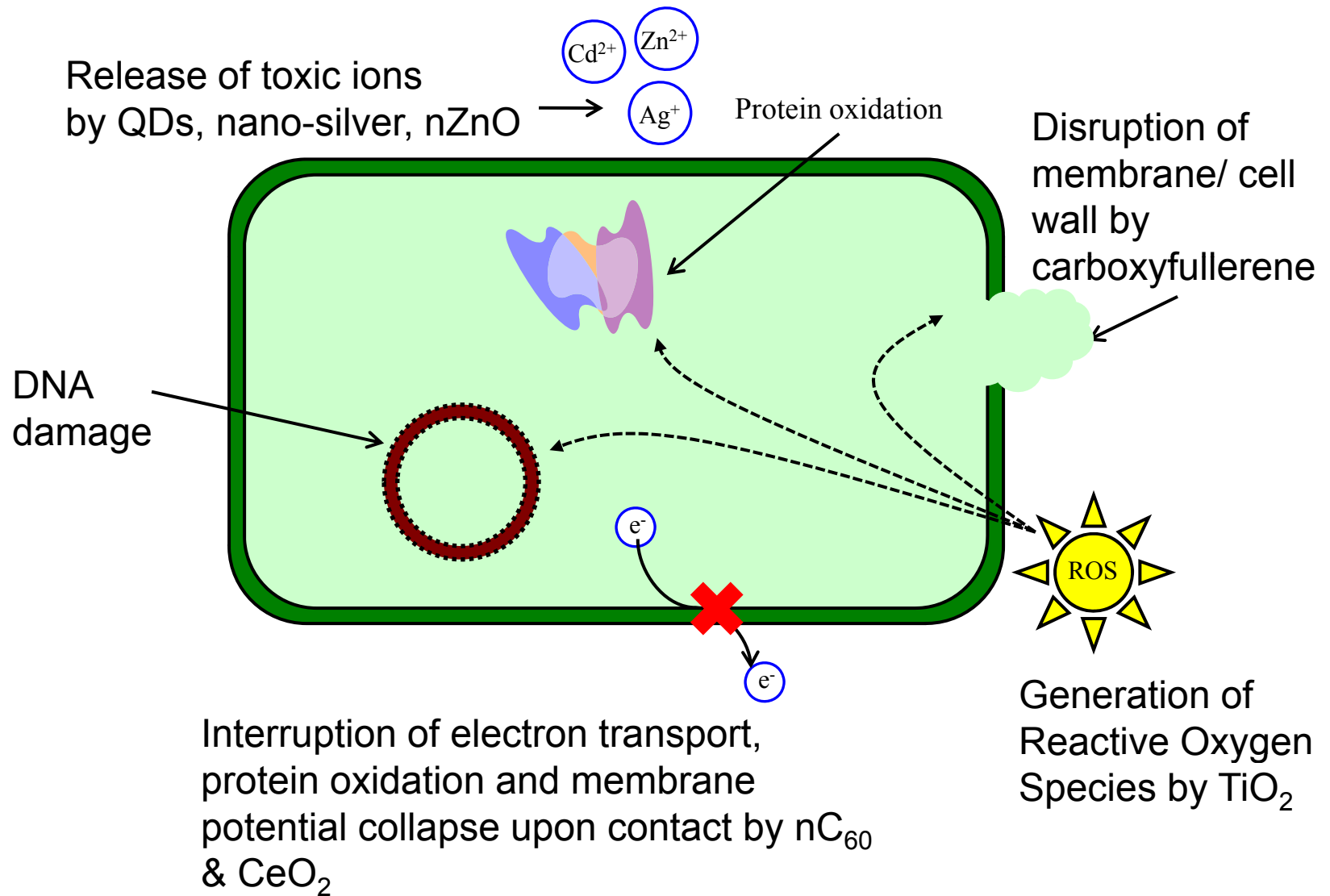
Decrease in *B. subtilis* membrane potential



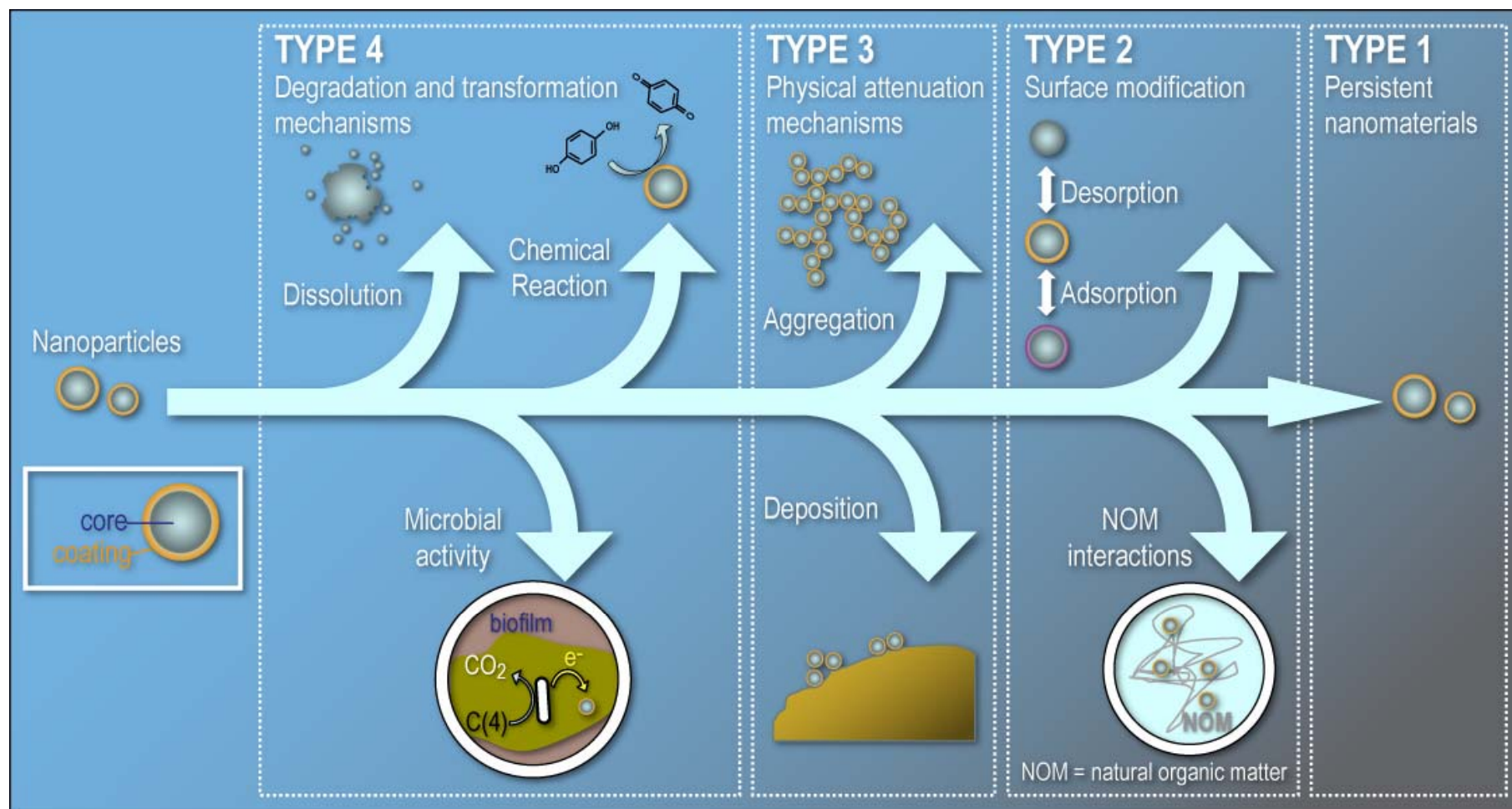
Decrease in *E. coli* respiration



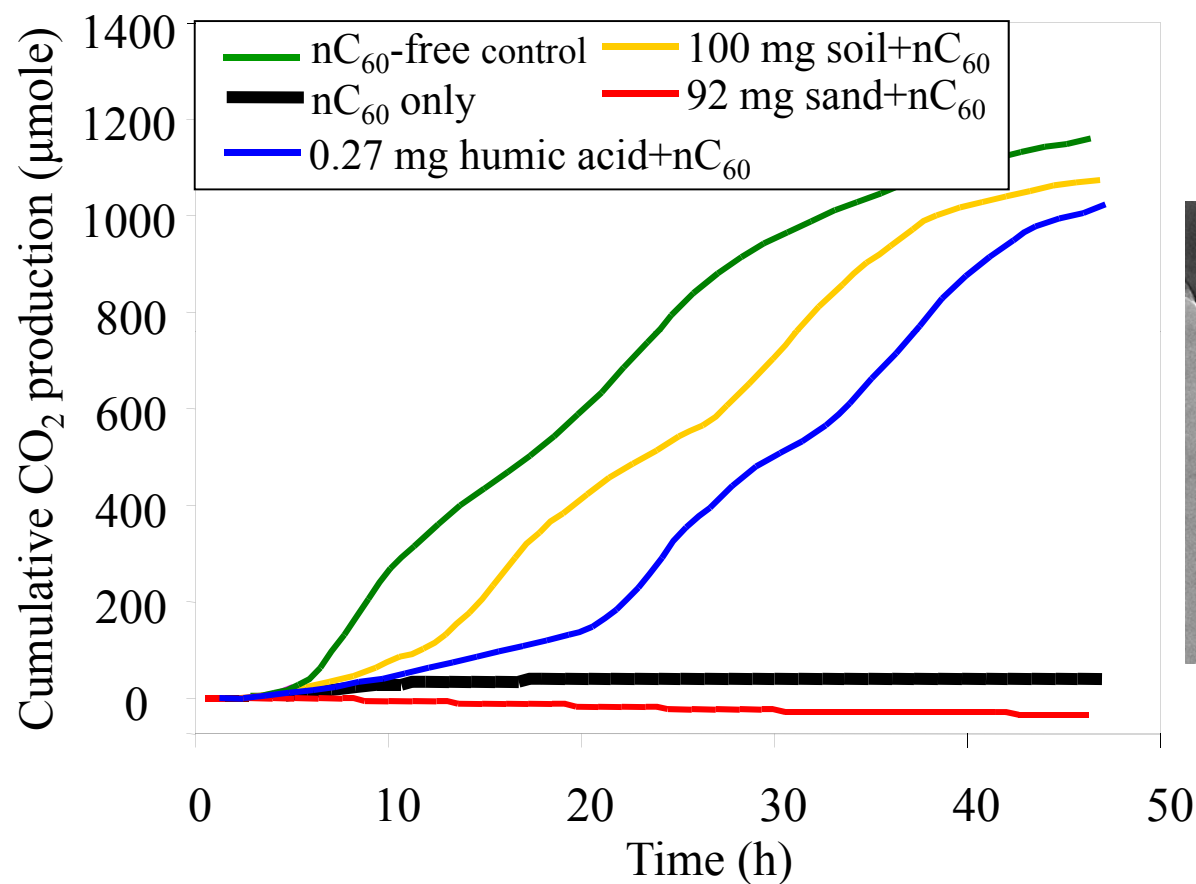
Bacterial Toxicity Mechanisms



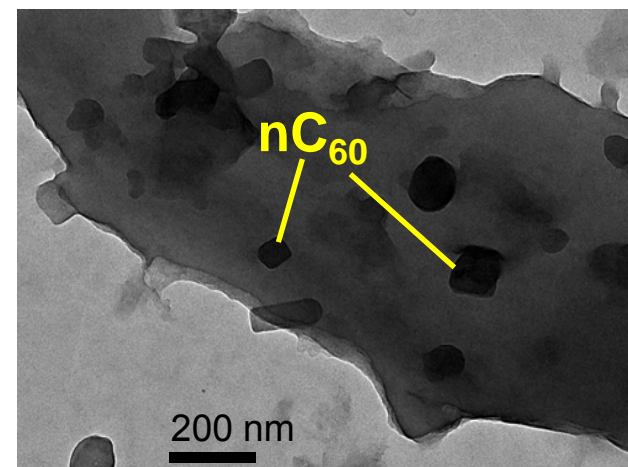
Nanoparticle Modifications in the Environment



NOM reduces bioavailability & toxicity of nC₆₀



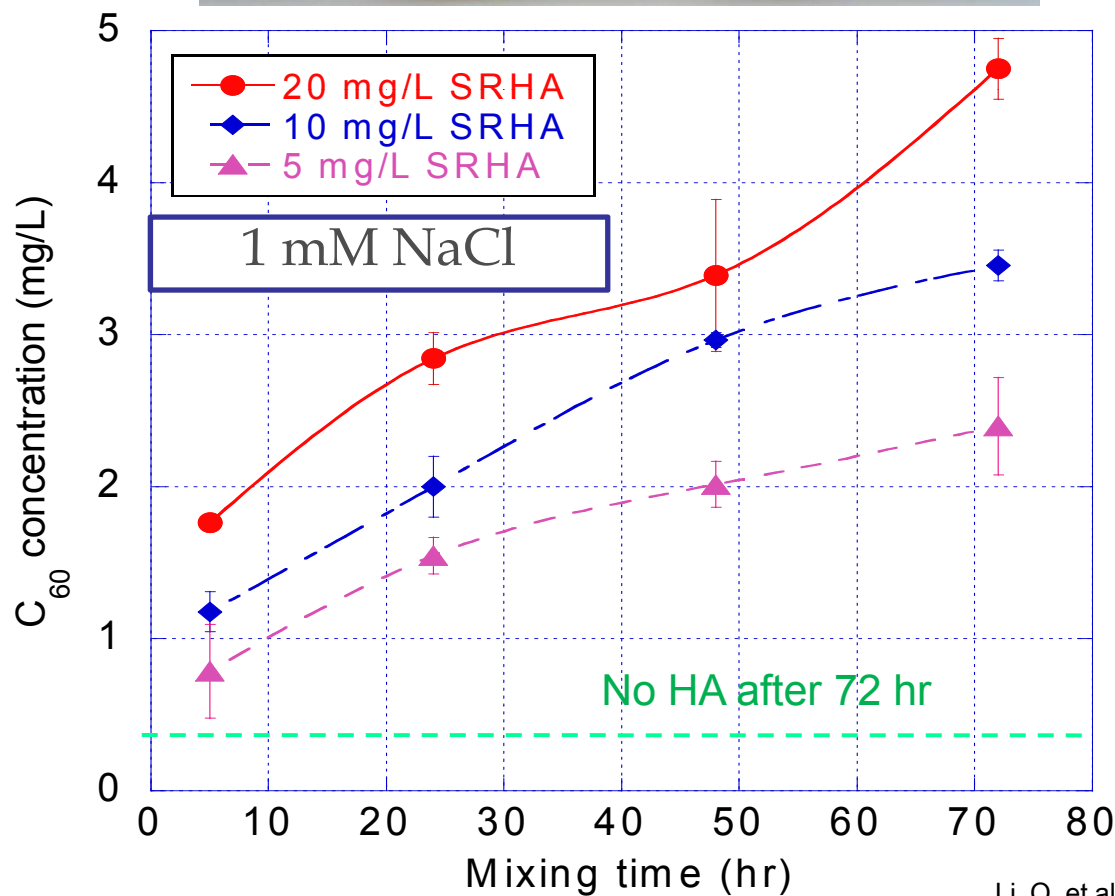
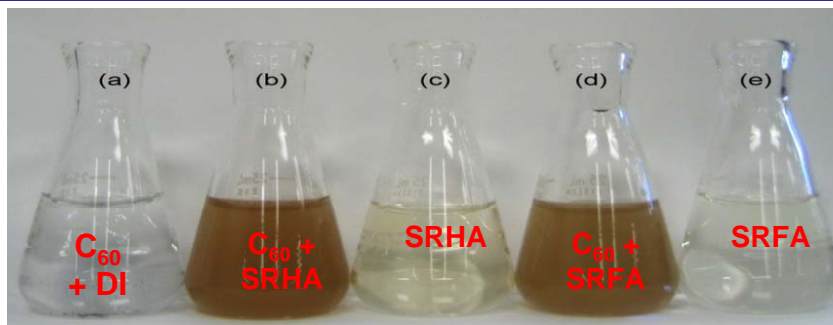
nC₆₀ trapped by humic colloids



Humic acid concentrations as low as 0.1 mg/L eliminated toxicity

Dissolved NOM Enhances C_{60} Dispersion

- Dispersed C_{60} was measured as dissolved TOC

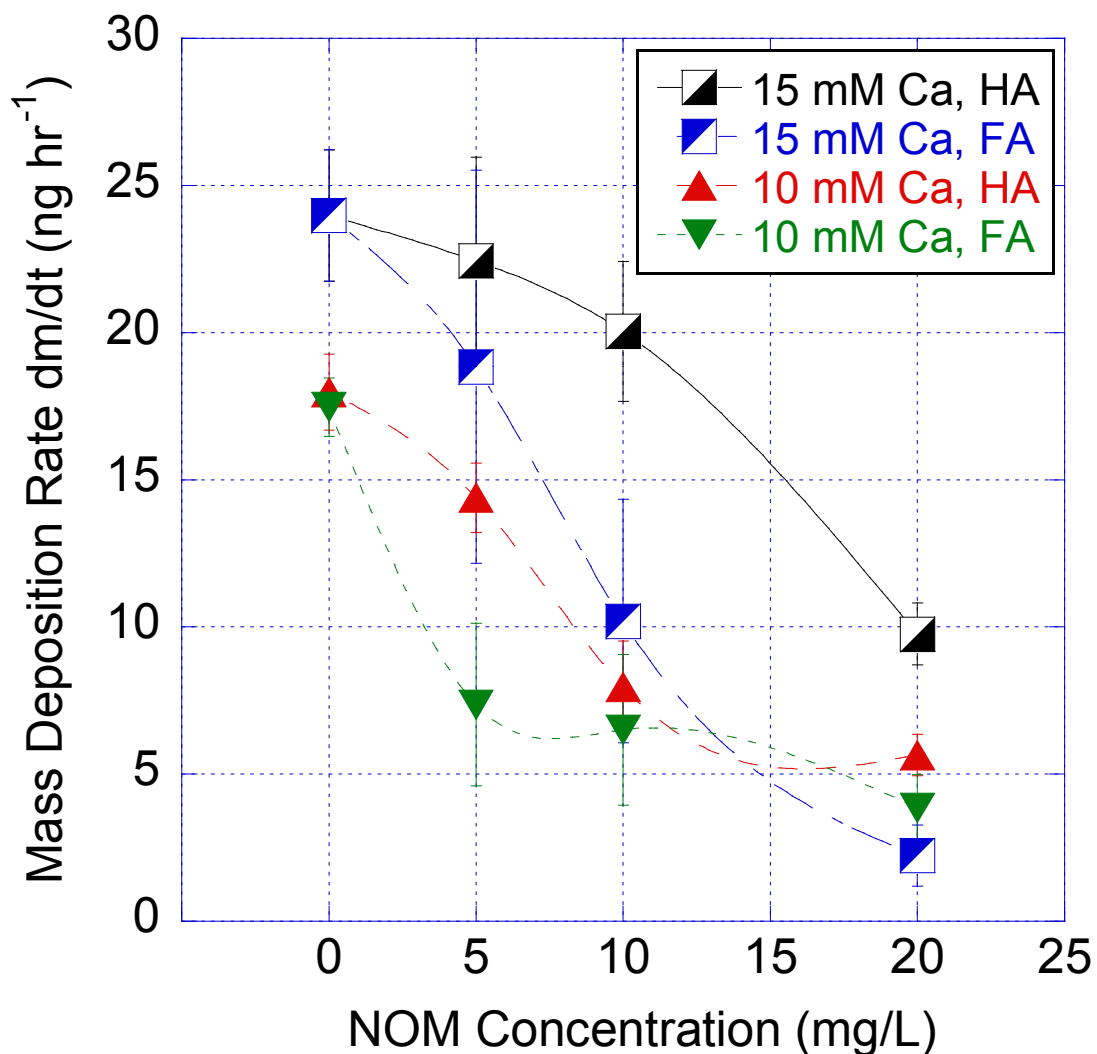
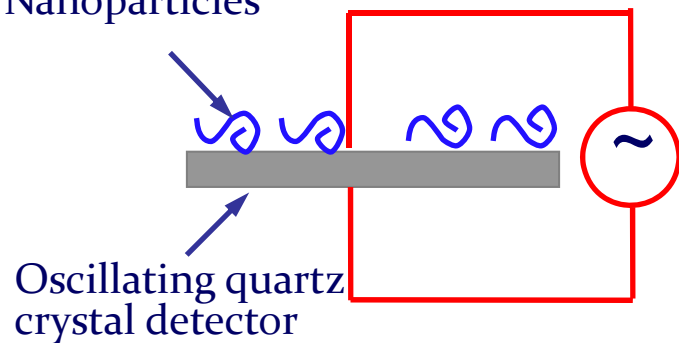


Dissolved NOM Decreases nC₆₀ Deposition onto a Quartz Surface, Increases Mobility in Water

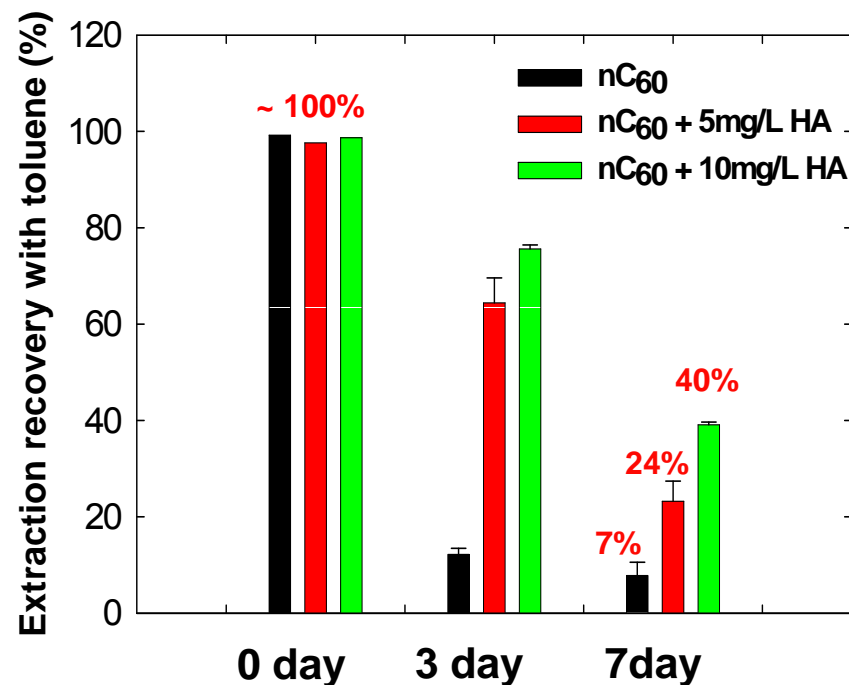
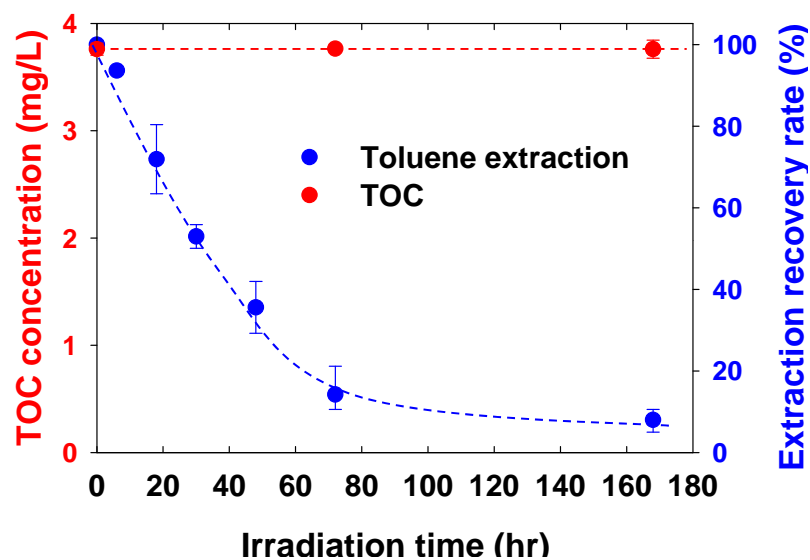
Quartz Crystal Micro Balance



Depositing Nanoparticles



NOM hinders photo-transformation of nC₆₀



In the presence of O₂ and UV light,
The surface of nC₆₀ is transformed
(C-OH, C=O, C=O).
This hinders its extraction with toluene

Dissolved humic and fulvic acids
hinder this transformation, enhance
extraction with toluene

Risk = Hazard × Exposure



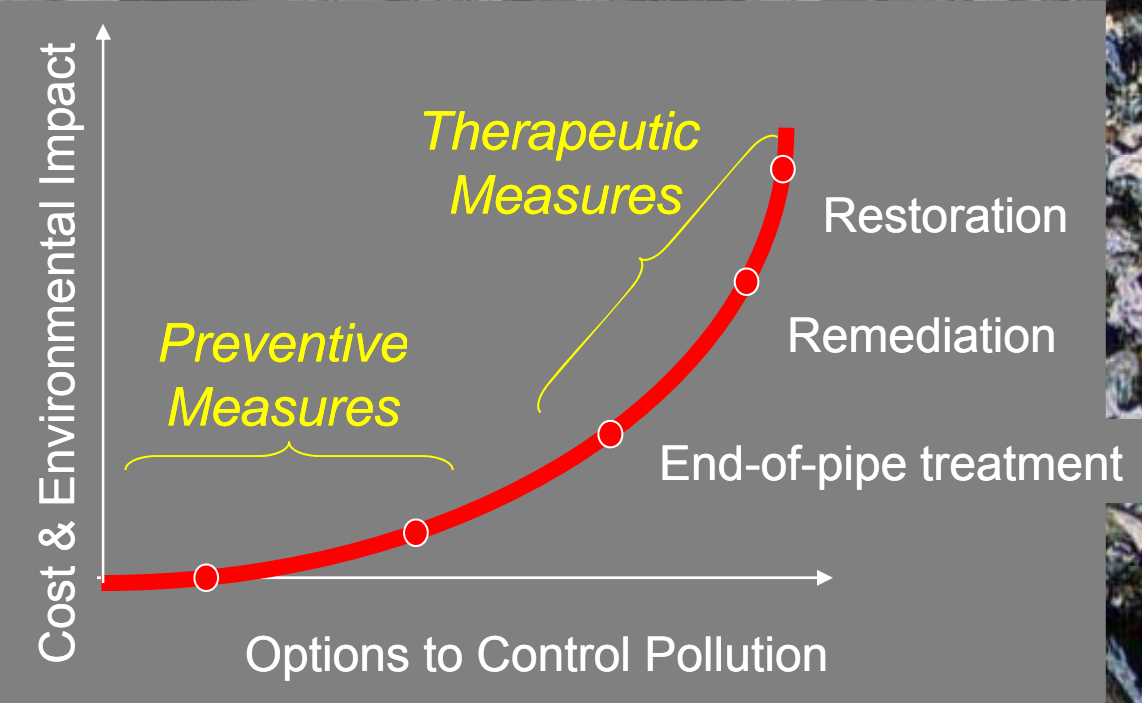
Hazard, but no exposure



Exposure but no hazard

Hazard as well as exposure





The 12 Principles of Green Nanotechnology

1. **Inherent rather than circumstantial** (use raw materials and elements that are inherently non-hazardous if released)
2. **Prevention rather than treatment** (containment, minimize exposure, *design away hazardous functionalities or features without impacting useful functions?*)
3. **Design for separation and purification of wastes** (take advantage of magnetic properties for separation? Stabilizing coatings that can be intentionally removed?)
4. **Maximize mass, energy, space and time efficiency** (multi-functionality of NPs may be desirable)
5. **“Out-pulled” rather than “input-pushed” through the use of energy and materials** (quality > quantity, need > greed, enough > more, long-term > short-term)

The 12 Principles (Continued)

6. Find opportunities for recycle, reuse or beneficial disposition (non toxic NPs that enhance nutrient or water retention and soil fertility?)
7. Target durability rather than immortality (pick the “right” coatings, avoid indefinite persistence)
8. Need rather than excess - don't design for unnecessary capacity – avoid “one size fits all” (incorporate just what you need, avoid excess MNMs in commercial products)
9. Minimize material diversity to strive for material unification and promote disassembly + value retention (take advantage of economy of scales, minimize variability and sources of a given MNM?)

The 12 Principles (Continued)

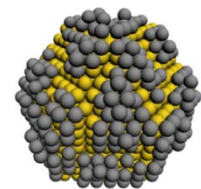
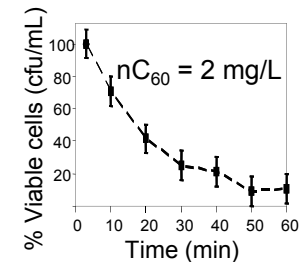
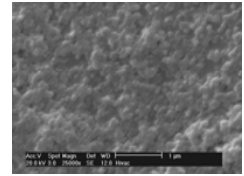
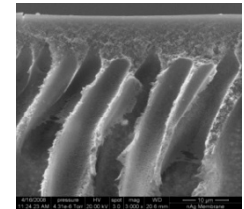
10. **Integrate local material and energy flows** (holistic LCA perspective, look for interconnectivity, system of systems)
11. **Design for commercial “afterlife”** (enable recycling, remanufacturing and/or reuse opportunities? beneficial disposition?)
12. **Use renewable & readily available inputs through life cycle** (minimize carbon, land use and water footprint?)

Potential Leapfrogging Opportunities for Microbial Control and Disinfection



Some Potential Applications

- Fouling-resistant membranes
 - Zodrow K., L. Brunet, S. Mahendra, Q. Li, and P.J.J. Álvarez (2009). Wat. Res. 43:715-723.
- Antimicrobial coatings
 - Lyon D.Y., D. Brown, E. Sundstrom, and P. J.J. Alvarez (2008). Int. Biodeterior. Biodegrad. 62:475-478.
- Chemical Disinfection without DBPs
 - Li Q., S. Mahendra, D. Y. Lyon, L. Brunet, M. V. Liga, D. Li and P. J.J. Alvarez (2008). Wat. Res. 42:4591-4602.
- Pd/Au Hypercatalysts for reductive dechlorination
 - Wong M.S., P.J.J. Alvarez, Y.L. Fang, N. Akçin, M.O. Nutt, and K.H. Heck (2008). J. Chem. Technol. Biotechnol. 84:158-166.



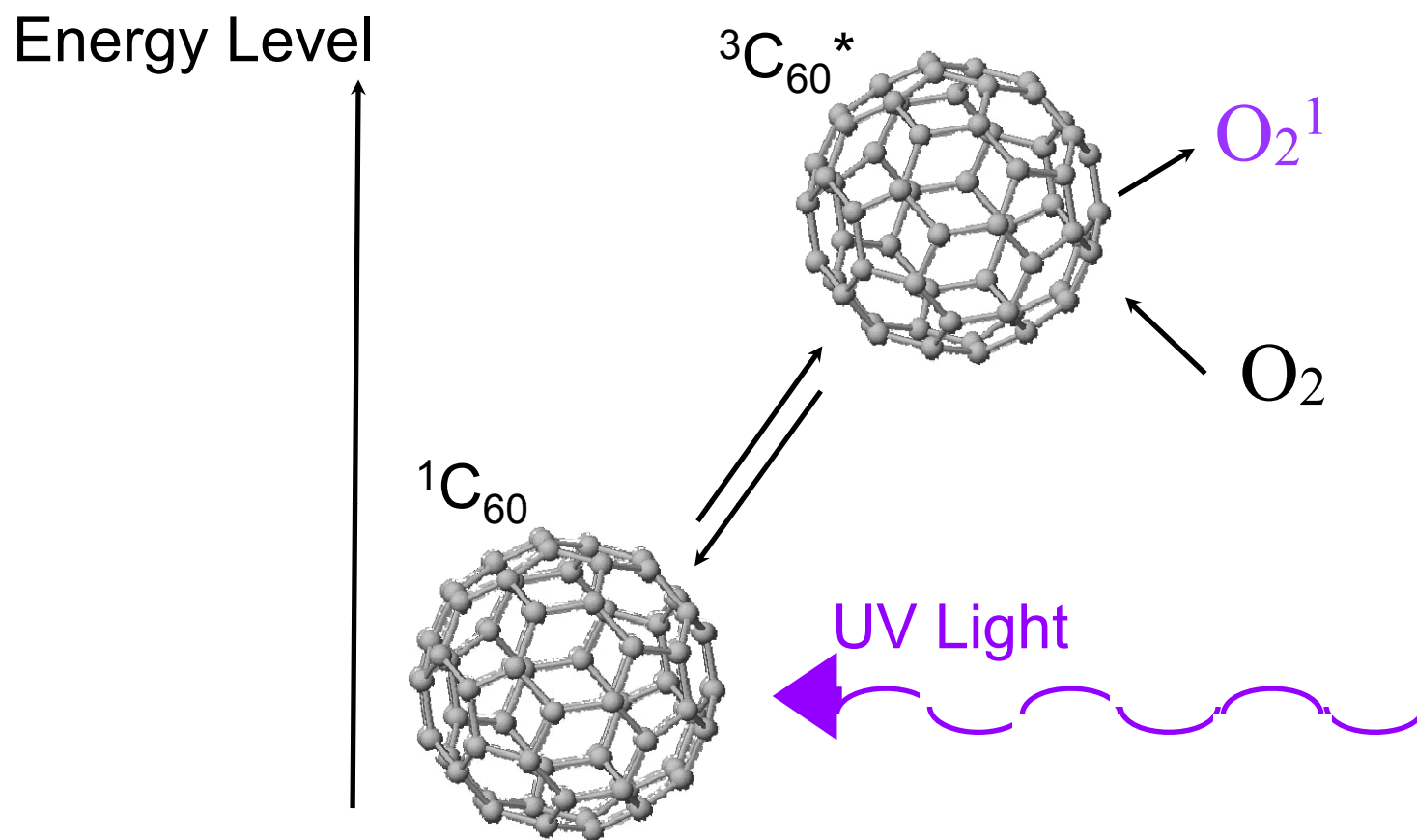
- **Photocatalytic Disinfection with fullerenes**



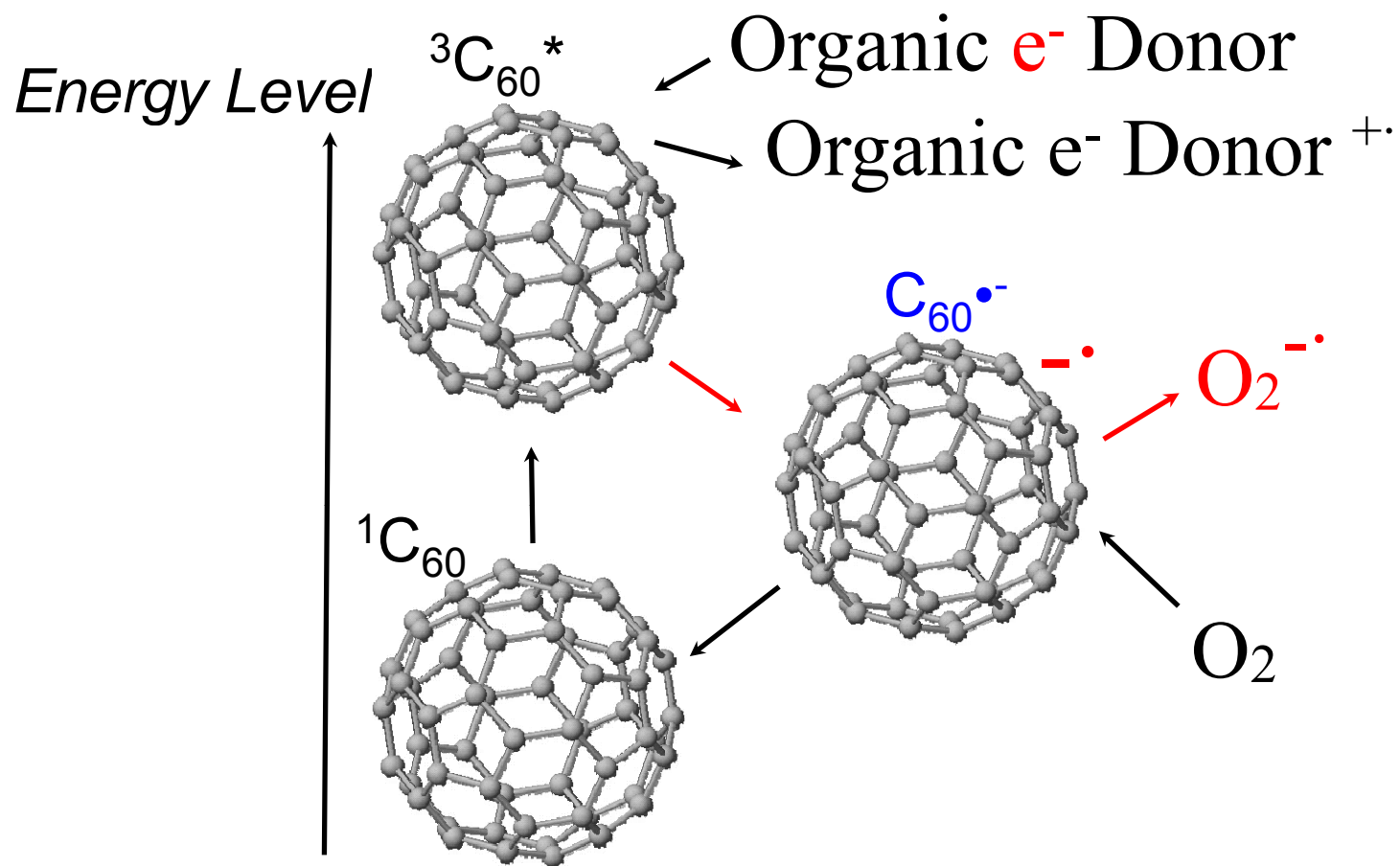
Potential Application: Enhancing UV Disinfection

- UV disinfection is increasingly used to inactivate cyst-forming protozoa such as *Giardia* and *Cryptosporidium*.
- However, UV is relatively ineffective to treat virus unless the contact time and energy output are significantly increased





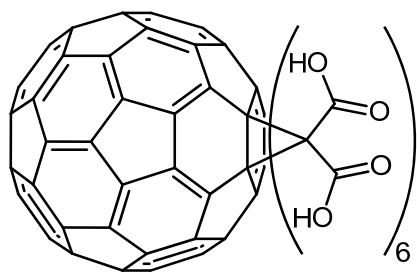
Light excites C_{60} to triplet state. Energy transfer between $^3C_{60}^*$ and molecular oxygen gives rise to singlet oxygen (1O_2)



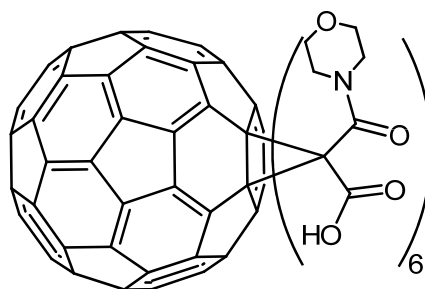
${}^3\text{C}_{60}^*$ transforms to C_{60} radical anion ($\text{C}_{60}^{\bullet-}$) in the presence of some electron donors, which reduces O_2 to **superoxide** ($\text{O}_2^{\bullet-}$)



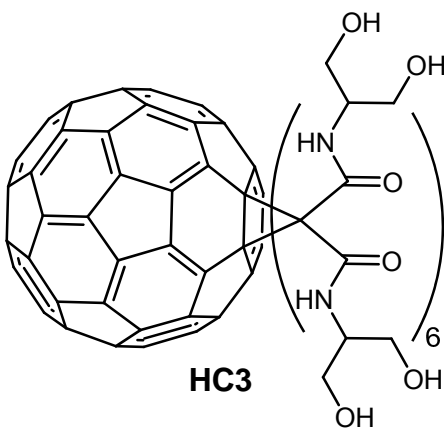
"Water Soluble" Derivatized Fullerenes



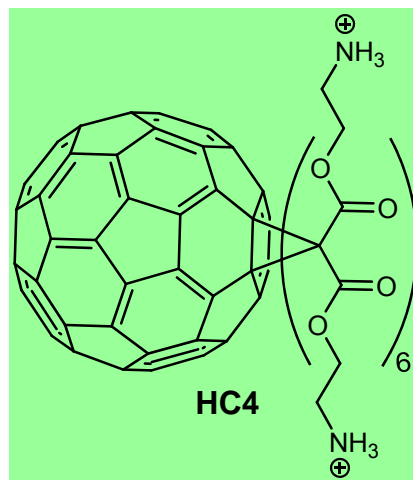
HC1



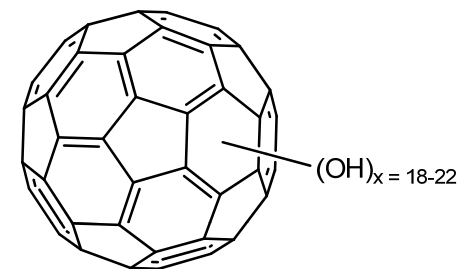
HC2



HC3



HC4



Fullerol

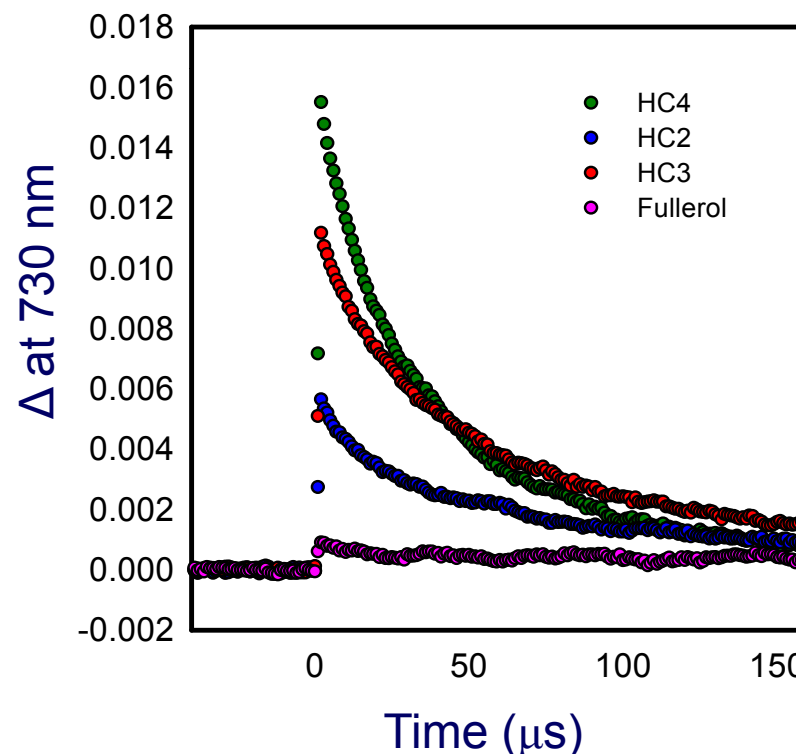
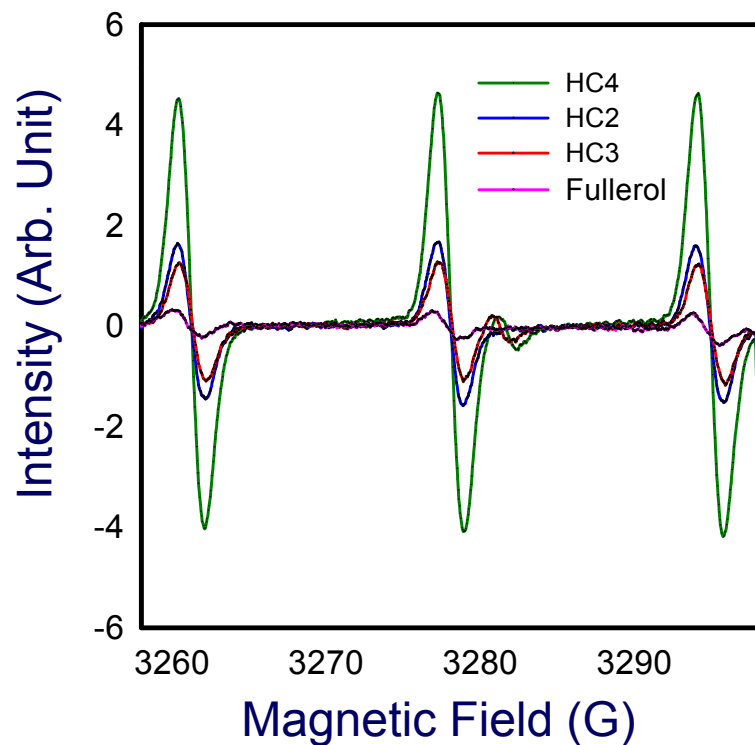
VS

* Commercially Available, MER Corp.

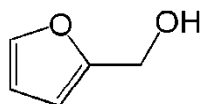
* Synthesized in Lon Wilson's lab, Dept of Chemistry, Rice University (Bingel Reaction)

Superior Photosensitized $^1\text{O}_2$ Production

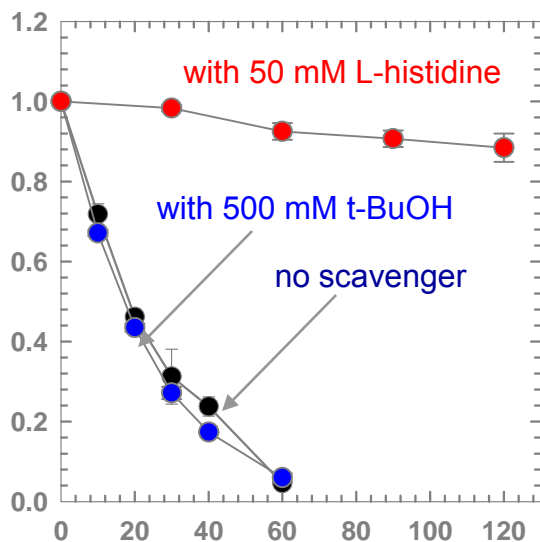
- Larger Electron Paramagnetic Resonance Spectra Peaks Correspond to Higher $^1\text{O}_2$ Generation (All Outperform Fullerol)
- Nanosecond Laser Flash Photolysis Confirms Long-lived Triplet State, Conducive to Efficient ROS Production (HC4 >> Fullerol)



Significant $^1\text{O}_2$ production despite clustering



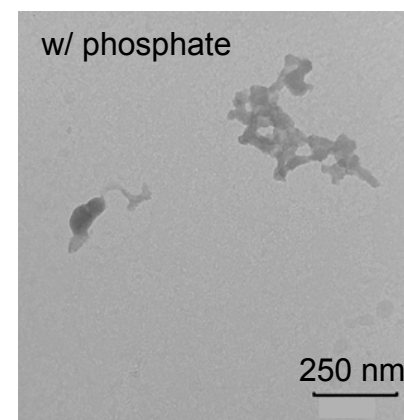
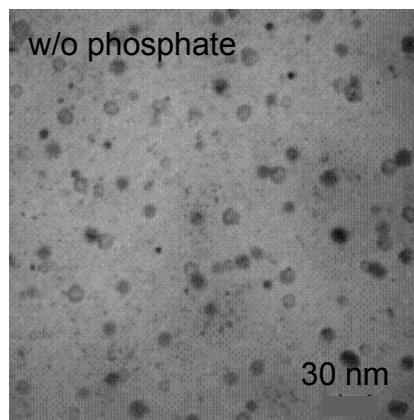
FURFURYL ALCOHOL
($^1\text{O}_2$ INDICATOR)



MIC test with *E. coli* shows minimum toxicity

DERIVATIVES	MIC (μM)	DIAMETER (nm)
HC1	> 400	1.32 / 1.12 ^a
HC2	> 400	281.1 / 60.5
HC4	120	1.50 / 1114
Fullerol	> 400	2.43 / 6.10

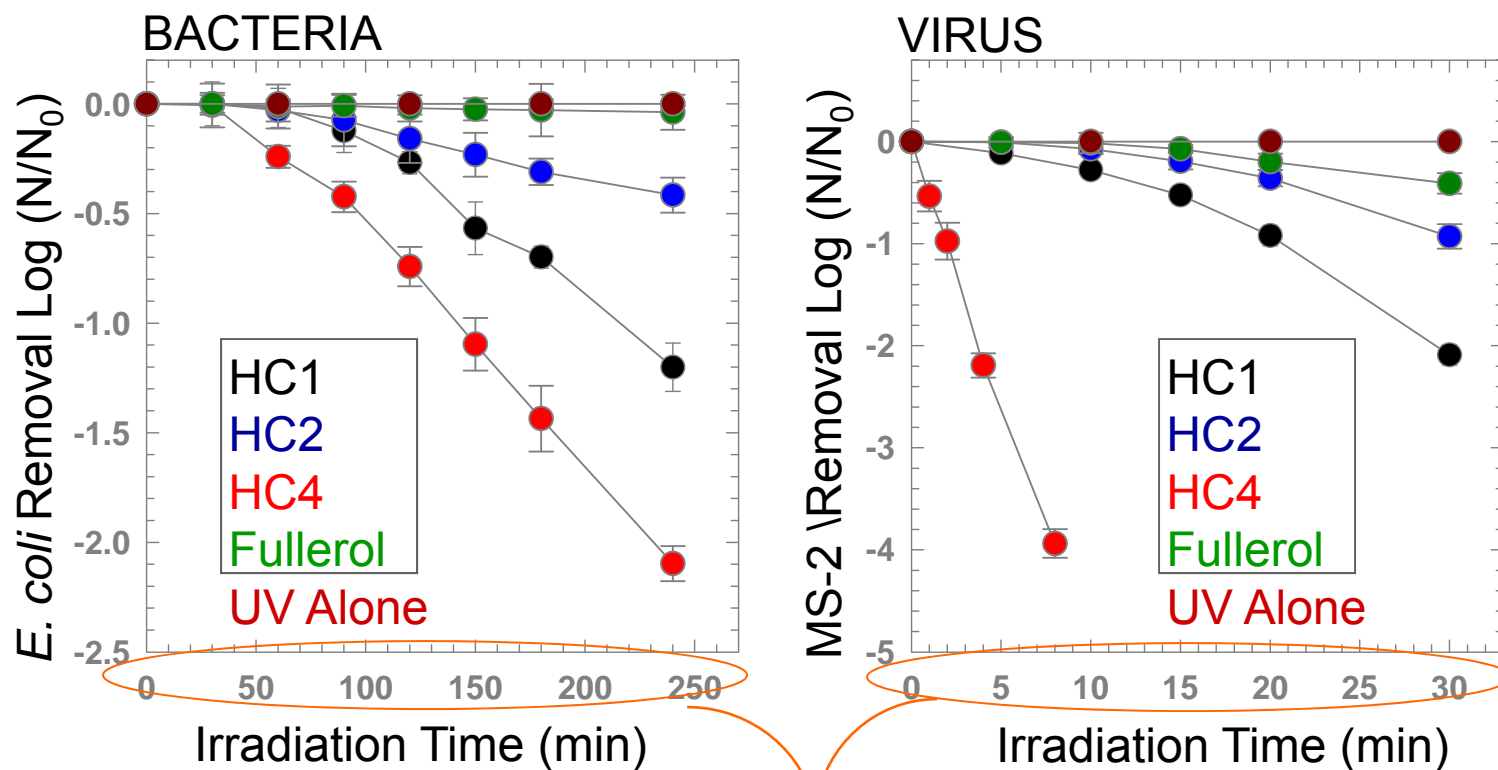
^a: determined in the presence of 10 mM phosphate





Photocatalytic Disinfection

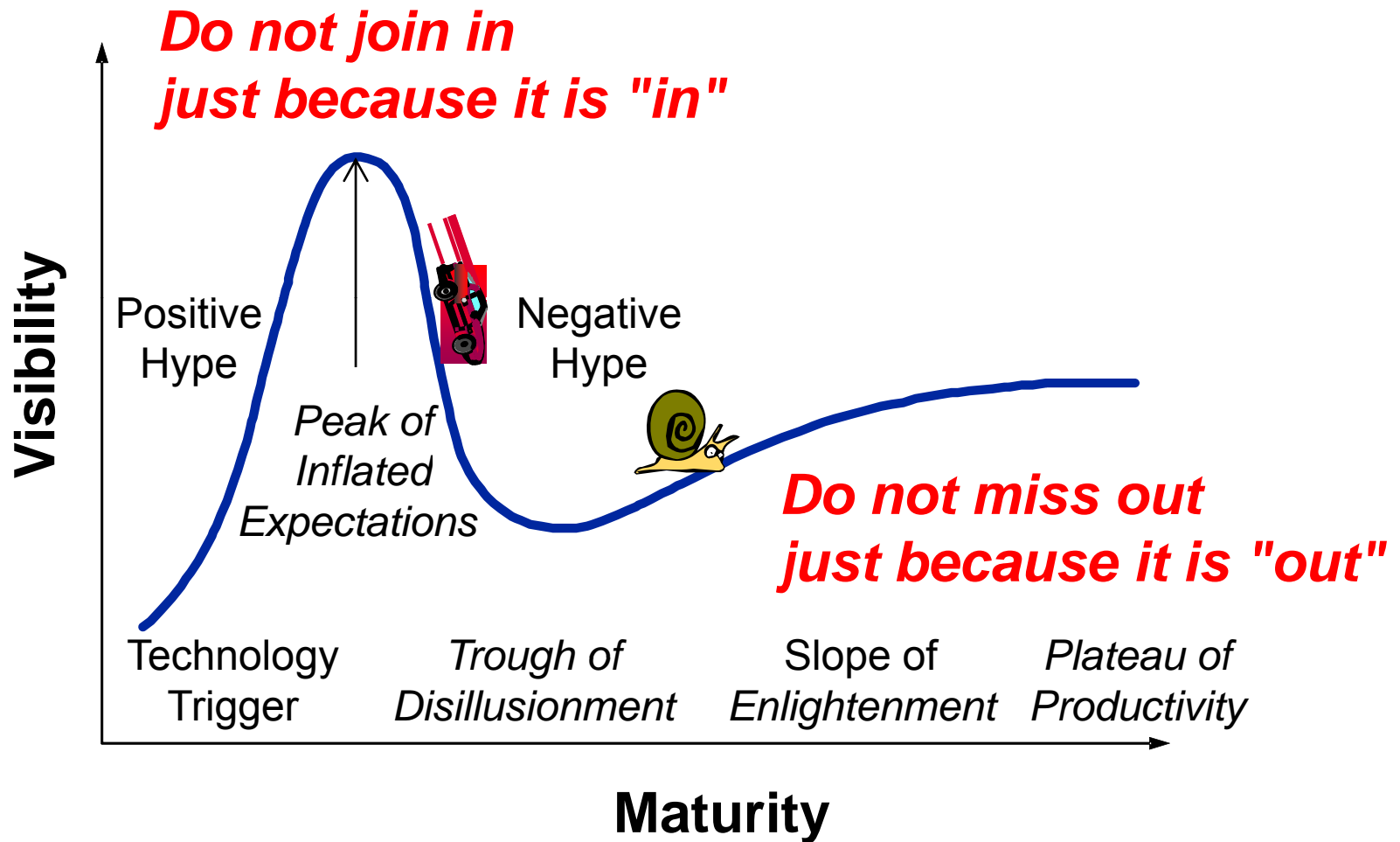
$[C_{60} \text{ derivative}] = 50 \mu\text{M}$, BLB lamp 350 - 400 nm, $I = 6 \times 10^{-6} \text{ Einstein} \cdot \text{min}^{-1} \text{L}^{-1}$



BETTER PERFORMANCE FOR VIRAL INACTIVATION
(unusual for traditional disinfection with UV, O₃, Cl₂ etc.)



Quo Vadis, Nano?



"Nanohype" - Berube

So What?

- Implications:
Ecotoxicology- Biodiversity
and food webs?
Biogeochemical cycling?
Mitigated by NOM, salts
- Applications:
DBP-free disinfection
(sunlight-driven?), Advanced
oxidation of micropollutants?
Antifouling or anticorrosion
coatings? Membranes?

