



# Direct and indirect toxic mechanism of nanoparticles. Evidence and models.

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## **Environmental emissions and exposure remain largely unknown!**

\* Mueller and Nowack, ES&T (2008)  
(models) in Switzerland surface water.

CNT : 6 Kg/year,

Nano-Ag : 620 kg/year,

NanoTiO<sub>2</sub> : 47300 Kg/year gives PEC of 0.016 mg/l.

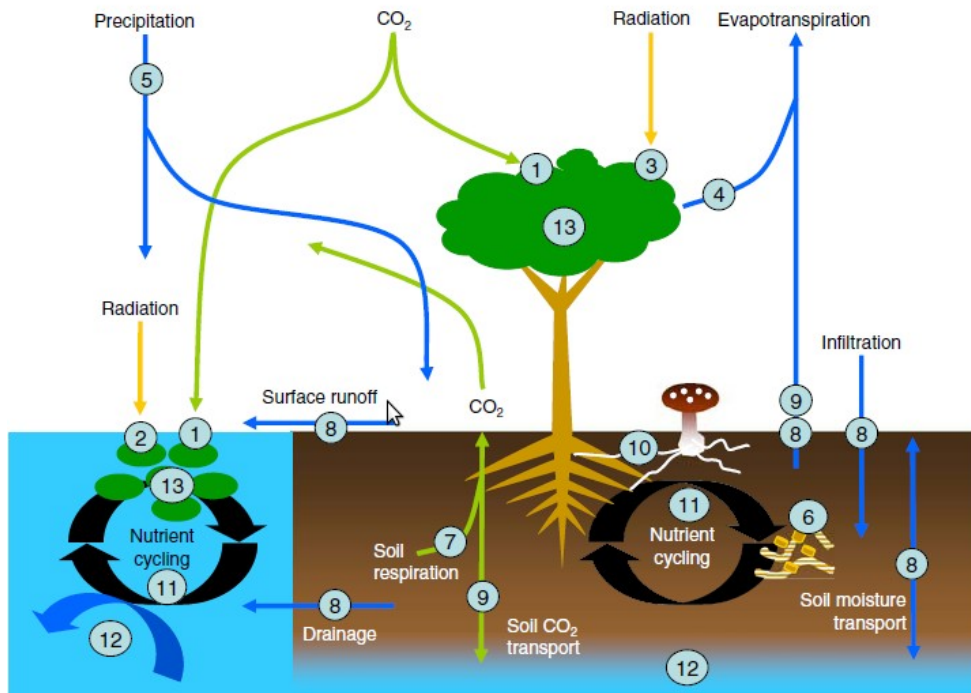
\* Park et al. (2008) gives for CeO<sub>2</sub> (exp. And models) 0.2 g/year (0.07 % of the PM exposure)

**Environmental accumulation** in sediments can be significant.

Koelmans, Nowack and Wiesner, Environ. Pollut. (2009)  
factors of 1000-10000 are realistic. Calculated up to  $4 \cdot 10^6$  for carbonaceous ENP.

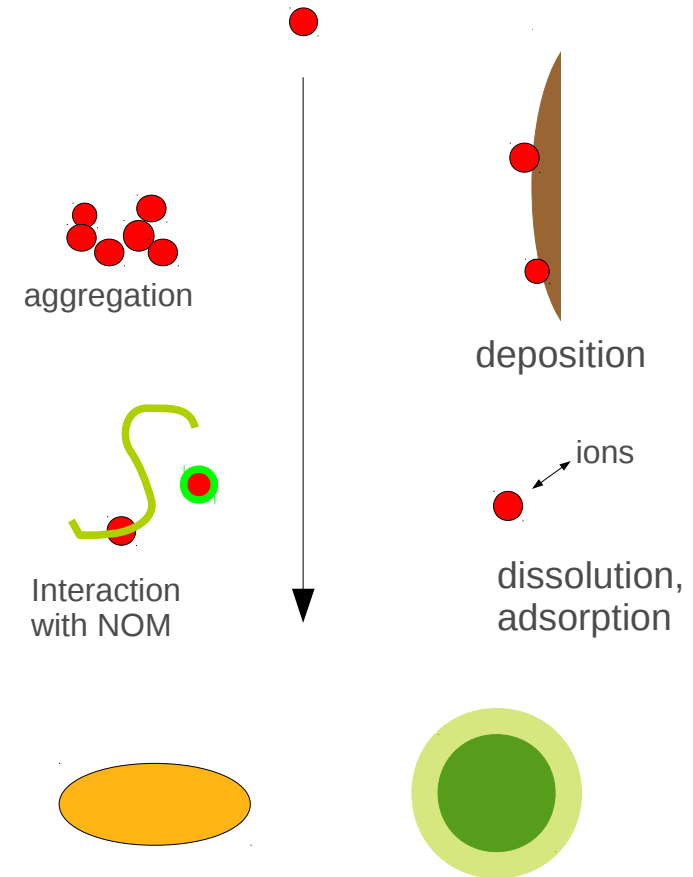
# Introduction

Simplified scheme of some terrestrial and aquatic ecosystem processes (Navarro et al., Ecotoxicology 2008 ).



NPs interact with a complex ecosystem containing multiple and diverse organisms (bacteria, algae, fungi, plants...).

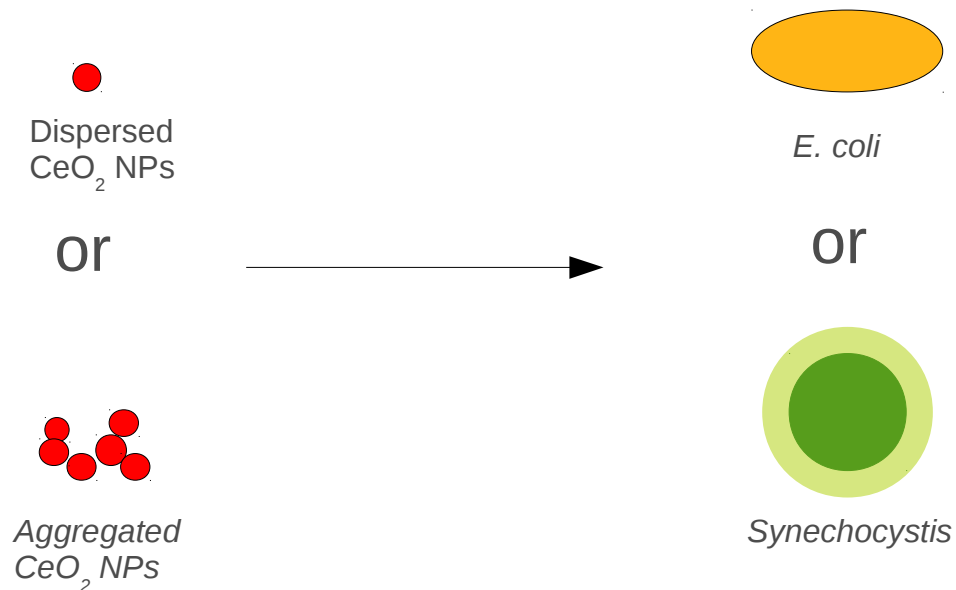
## NPs sources



**Modified NPs** interacts with organisms

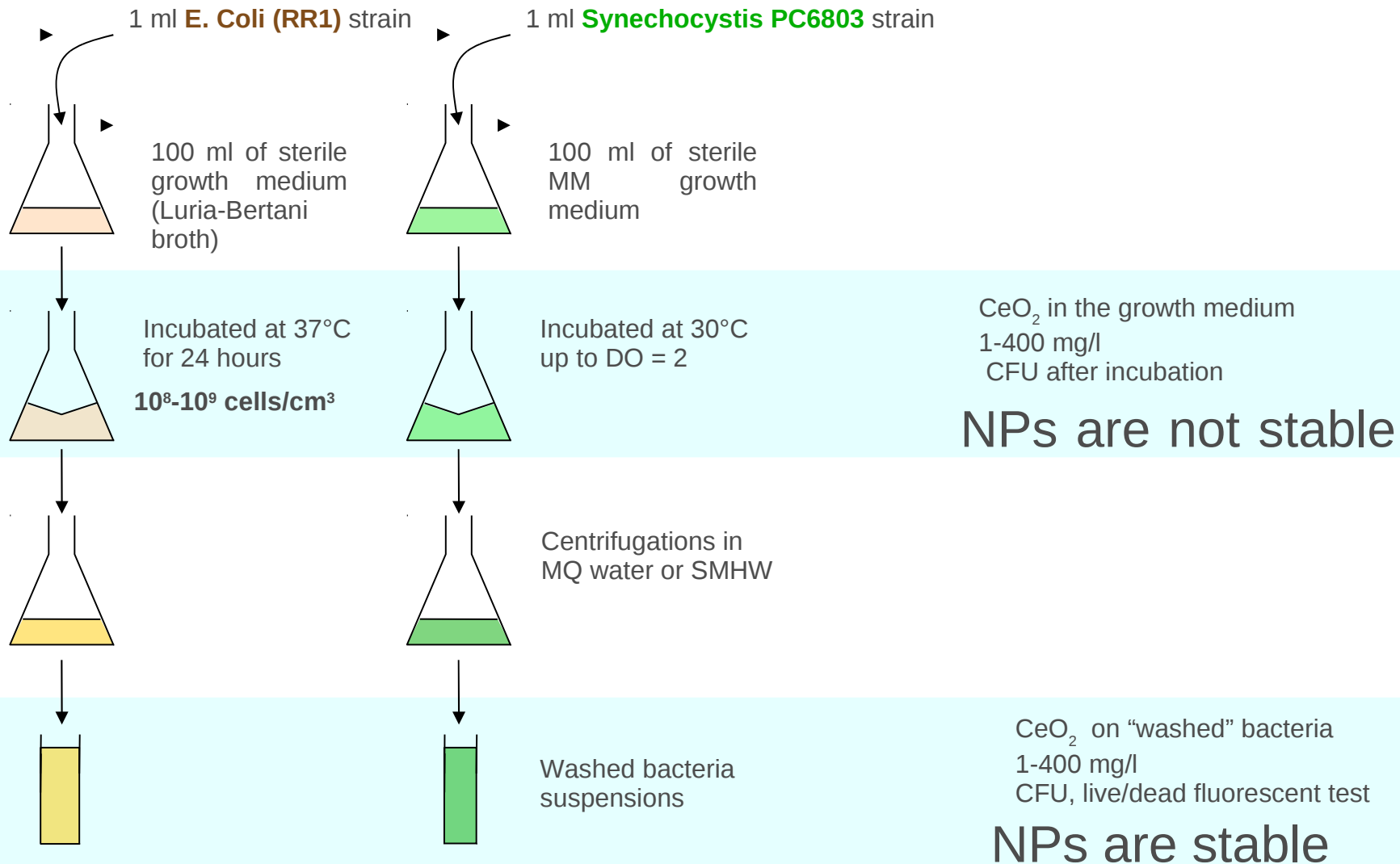
# Introduction

Interaction of aggregated and dispersed  $\text{CeO}_2$  NPs on *E. coli* or *Synechocystis*.

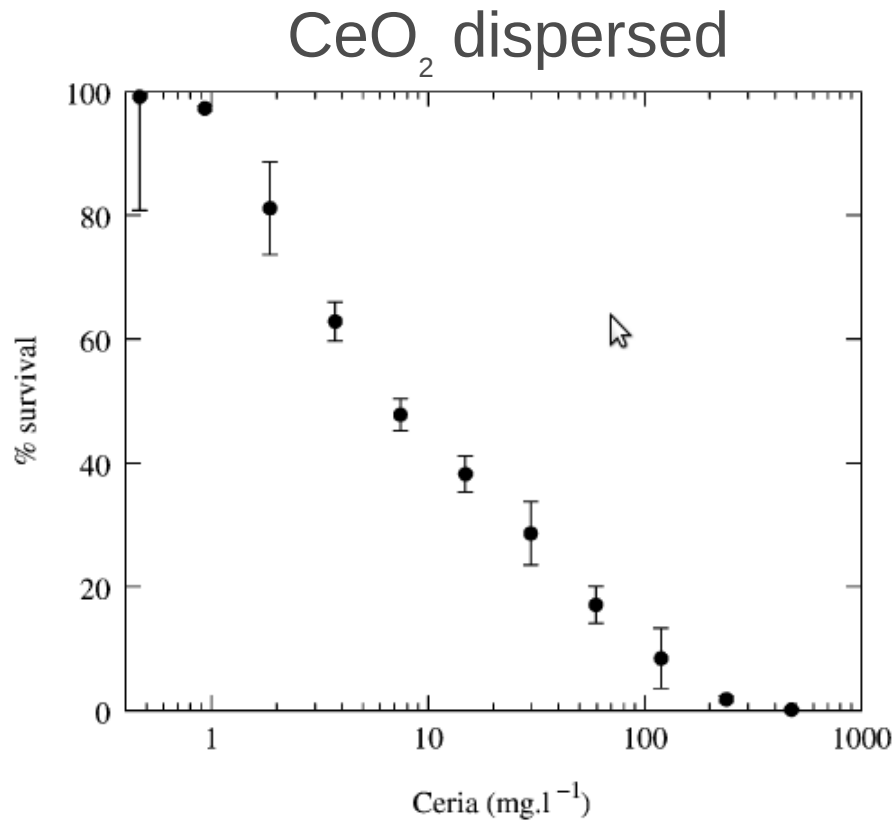


What is the toxicity and the mechanisms?

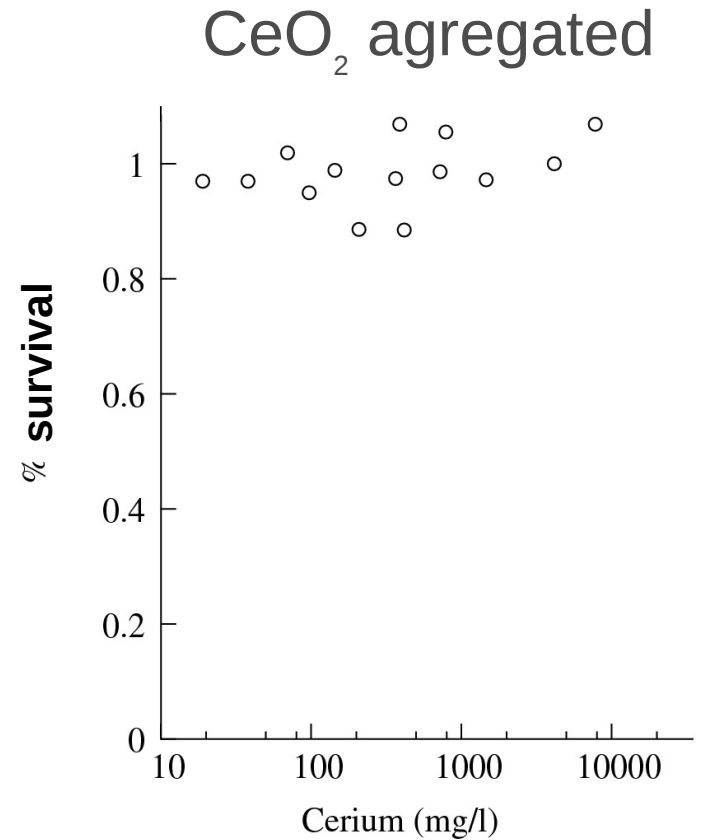
# Contact protocols



# Interaction between CeO<sub>2</sub> NPs and *E. coli*



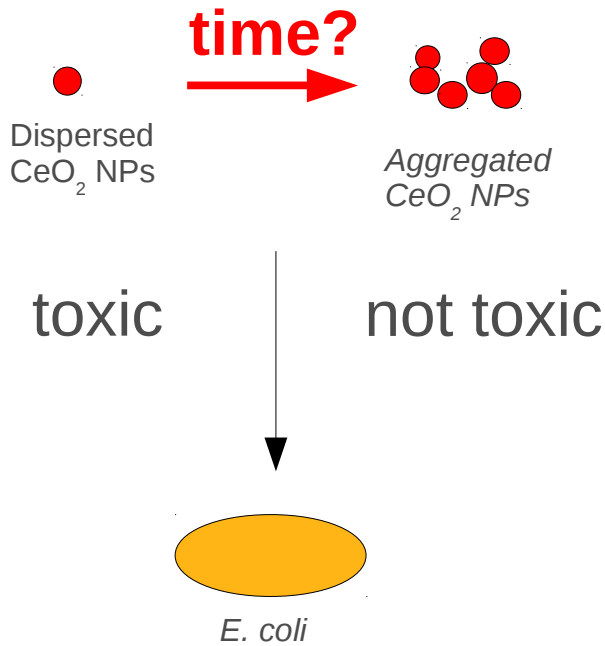
Strong adsorption at the bacteria cell wall.  
Clear dose response  
LD50 < 10mg/l.



Strong reduction of the CeO<sub>2</sub> NP. No toxicity.

Thill et al., ES&T 2006, Zeyons et al. Nanotoxicology accepted

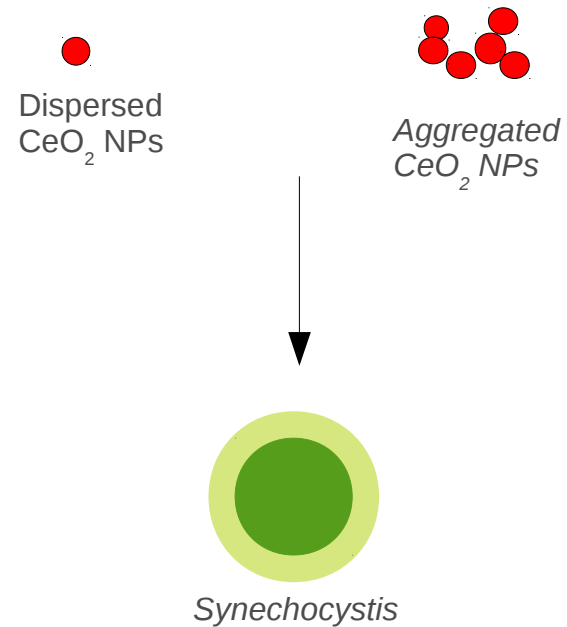
# Two mechanisms



Close contact between NPs and cells required.  
Only dispersed NPs are toxic

Thill et al. ES&T (2006),  $\text{CeO}_2$   
Lyon et Alvarez (2008), nC60  
Battin et al. ES&T (2009),  $\text{TiO}_2$

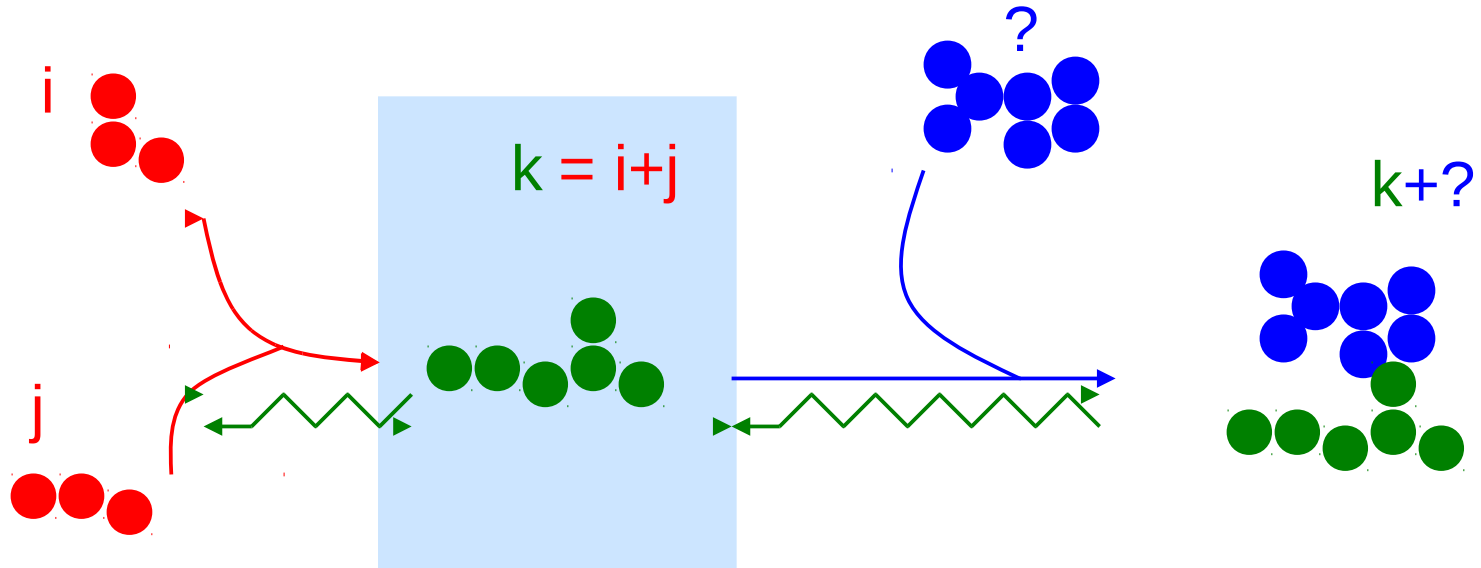
## "Direct" mechanism



Aggregation state is not important  
pH shift when not in buffered media controls the toxicity.  
Same category as : release of soluble toxic ions ( $\text{Ag}^+$ ,  $\text{Zn}^{2+}$ ,  $\text{Cd}^{2+}$ ), release of adsorbed toxic molecules etc...

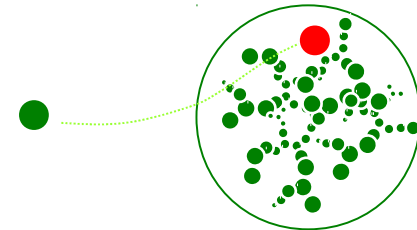
## "Indirect" mechanism

# Adsorption / aggregation kinetic model



$$\frac{dn_k(t)}{dt} = \sum_{i+j=k} \beta(i, j) n_i(t) n_j(t) - \sum_i \beta(i, k) n_i(t) n_k(t) + \sum_i F(i, k) n_{(i+k)}(t) - \sum_{i+j=k} F(i, j) n_{(i+j)}(t)$$

$\beta$  computed using collision kernels adapted for fractal aggregates. (Veerapaneni et Wiesner 1996)  
 Possibility of varying fractal dimensions during aggregation. (Thill, Wiesner et al. JCIS, 2001)



Adapted for nano-micro mixtures (work in progress CEINT, iCEINT)

# Time scale for NP aggregation ?

If only one nanoparticle type of size  $r_{NP}$  and concentration  $C$ .

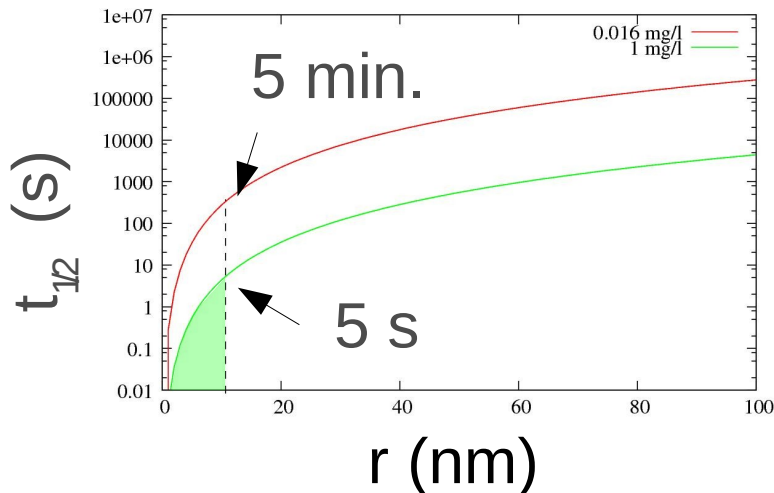
Then at  $t=0$  :  $n_1(0) = C/v_{NP} \rho$  and  $n_i(0)=0 \quad \forall i \neq 1$

$$\frac{dn_k(t)}{dt} = \sum_{i+j=k} \beta(i,j)n_i(t)n_j(t) - \sum_i \beta(i,k)n_i(t)n_k(t) + \sum_i F(i,k)n_{(i+k)}(t) - \sum_{i+j=k} F(i,j)n_{(i+j)}(t)$$

Equation reduces to :  $\frac{dn_1(t)}{dt} = -\beta(1,1)n_1(t)^2 \longrightarrow \frac{1}{n_1(0)} - \frac{1}{n_1(t)} = -\beta(1,1)t$

If  $t_{1/2}$  is the time when  $n_1(t_{1/2}) = n_1(0)/2$

$$t_{1/2} = \frac{\pi \rho v r_{NP}^3}{kT C}$$



$t_{1/2} < 5s$  for 10 nm NP at 1 mg/l

with  $v = 0.00089 \text{ kg}/(\text{m}\cdot\text{s})$

$T = 298 \text{ K}$

$\rho = 6500 \text{ kg}/\text{m}^3$

$C = 0.016 \text{ or } 1 \text{ mg}/\text{l}$

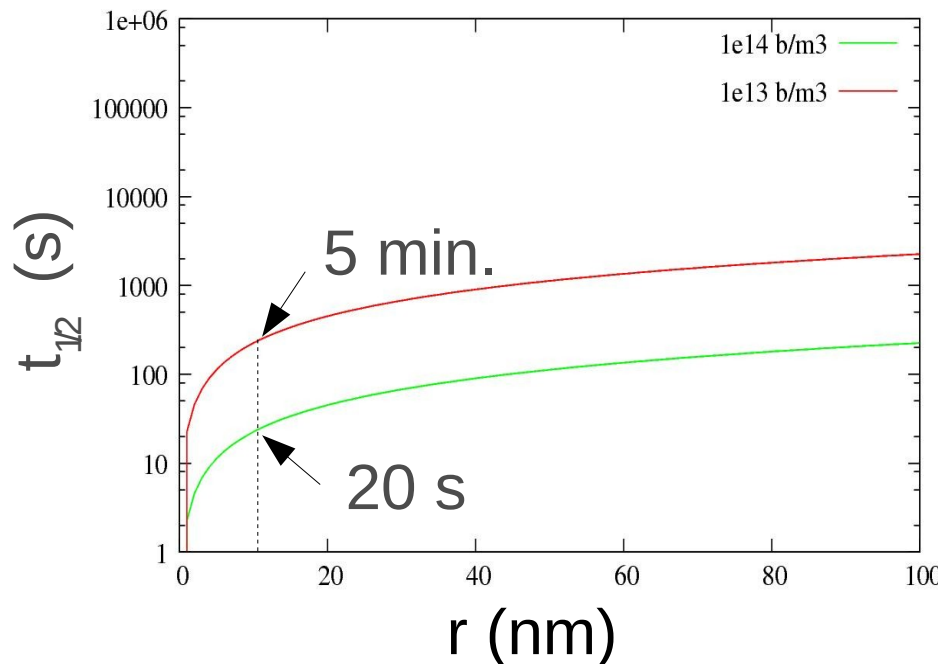
# Time scale for NP/bacteria adsorption ?

A mixture of NP at concentration  $n_s$  radius  $r_s$  and bacteria at a concentration  $n_b$  of radius  $r_b$ . At initial time, if only adsorption on bacteria is possible :

$$\frac{dn_s(t)}{dt} = -\beta(s,b)n_s(t)n_b(t) \approx \frac{2kT}{3\nu} \frac{r_b}{r_s} n_b(t)n_s(t) \quad \ln\left(\frac{n_s(t)}{n_s(0)}\right) = \frac{-2kT}{3\nu} \frac{r_b}{r_s} n_b t$$

If  $t_{1/2}$  is the time when  $n_1(t_{1/2}) = n_1(0)/2$

$$t_{1/2} = \frac{3\nu \ln(2)}{2kT} \frac{r_s}{r_b} \frac{1}{n_b}$$



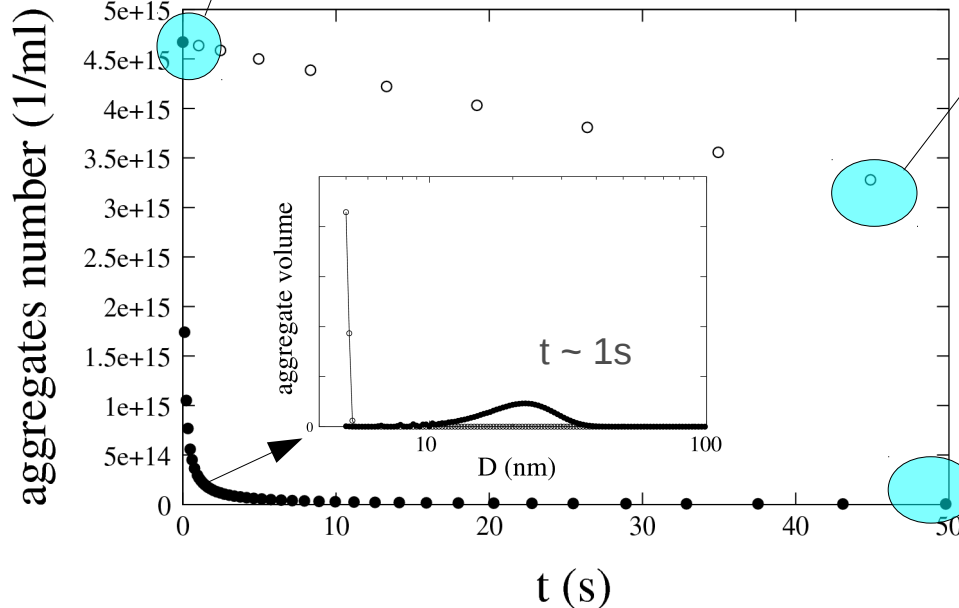
↑ Environmental concentrations

↓ Concentrated culture, biofilm

# Adsorption / aggregation kinetic model

Initial conditions:  
3 mg/l NPs  
 $10^8$  cells/ml

In water : stable NPs  
only adsorption



In medium : unstable NPs  
adsorption + aggregation

# Conclusions and perspectives

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- Toxicity of NPs can be induced by "**direct**" (requiring physical contact) or "**indirect**" mechanism.
- NPs aggregation state may not be a critical parameter for "**indirect**" toxicity.
- Competition between aggregation/deposition and adsorption has to be considered to assess potential "**direct**" toxicity especially in environmental relevant scenario.
- Are environmentally relevant exposure scenario in favor of "**direct**" toxic mechanism? (bio-accumulation? and genotoxicity!)

## Safe production and use of nanomaterials



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